

**Introduction to General, Organic and Biochemistry I**  
**Spring 2019 ( CHEM-30A-01 & 02 ) Syllabus**

**Lecture (Sections 01 & 02):** Monday & Wednesday -2:30 PM - 4:20 PM : Room S35

**Lab (Section-01):** Monday- 11:30 AM - 2:20 PM - Room SC2204

**Lab (Section-02):** Wednesday- 11:30 AM - 2:20 PM - Room SC2204

**Instructor:** Dr. Hema Ramakrishna

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**Office Hours:** Monday & Wednesday- 10:20 AM -11:20 AM SC1 second floor.

**Description:** An introduction General Chemistry course for Allied Health Fields with Laboratory. The topics covered in this course includes discussion of various measurement tools, energy and matter, discovery of an atom, introduction to elements, compounds, and types of bonding in compounds followed by various types of chemical reactions and stoichiometric calculations based on chemical equations. Properties of gases and solutions will be discussed. The course concludes with a discussion of acid-base chemistry and nuclear chemistry.

**Lecture text:** General, Organic and Biological Chemistry, Janice G. Smith, 4th Edn. McGraw-Hill.

**Lab Experiments:** The lab experiments are located at this link:

<https://www.deanza.edu/chemistry/Chem30A.html>

**Evaluation:** Your grade will be based on your performance in the following:

**Quizzes:** Ten quizzes will be given during class on Monday or Wednesday as scheduled in syllabus, and will have a time limit. If you miss the quiz, you will not have a chance to make it up. The best 9 quiz scores will be used in determining your final grade. Each quiz counts for 10 points.

**Exams:** There will be three exams and one final exam. You are permitted to bring a molecular model kit, the instructor must approve if it is assembled in any way. A scientific calculator that has at least log and exponential functions is required. **NO GRAPHING CALCULATORS.** Once the exam begins you may not leave the room unless you turn in the exam. The best 2 exam scores will be used in determining your final grade. Each exam counts for 100 points.

**No cell phones during exams and quizzes.**

Make-up exam shall be given for serious and compelling reasons only. Consult your instructor **PRIOR TO EXAM TIME** by all means. There will be 10% deduction in grades for all the make-up exams. If you feel that any of your exams are graded incorrectly, you are always welcome to **turn the exam in for a complete re-grade at the end of the lecture or laboratory period on the same day the exam is passed back.**

**Final Exam:** A comprehensive final exam will be given. Student who miss or fail the final exam will not receive a grade C or better.

**Labs:** All 7 labs count towards your grade. No make-up labs. Late labs will incur a penalty. You **MUST** wear eye protection during lab. Maintaining Lab safety is a primary concern, it is important to understand and follow the safety rules provided later in this syllabus.

**Lab final Exam:** There is one lab exam worth 70 points. The lab exam will be given during your regularly assigned laboratory sessions. **No early, late or make-up lab exams will be given and lab exam scores will count toward your overall course grade.**

9 Best Quizzes (10 pts each, 1 lowest quiz score will be dropped)	90 points
7 Lab reports (20 pts)(5 pts for prelab and 15 pts for report)	140 points
1 Lab Final	70 points
2 Best Exams (100 pts each, 1 lowest exam score will be dropped)	200 points
1 Final (200 pts)	200 points
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Total	700 points

Letter grades will be assigned according to the following grade scale:

90-100% = A
87-89.9% = A-
84-86.9% = B+
79-83.9% = B
76 -78.9% = B-
72-75.9% = C+
60-71.9% = C
50-59.9% = D
Below 50% = F

Dr. Ramakrishna reserves the right to change exam dates as well as modify the grade scale at any point during the quarter.

**Lab Notebooks:** You are required to maintain a bound laboratory notebook. Each experimental procedure must be written in your lab notebook prior to performing the experiment. All the observations and results must be entered immediately and directly into the lab notebook using pen. Lab final exam will be “**open-notebook**”. A well prepared notebook will be helpful during the exam.

**Lab Reports:** All Lab reports must be typed. Instructions to submit lab reports will be given during the lab sessions.

**Attendance:** Your attendance is urged for all lectures and required for all quizzes, exams and labs. Unexcused exam, quiz and lab absences score zero. It is the responsibility of the student to contact the instructor regarding missed work. If an absence is anticipated, the student should make arrangements to complete the missed assignments prior to the absence. In an emergency, it is the student’s responsibility to contact the instructor within one class period of an exam. There are no laboratory make-up days. If you miss lecture, laboratory lecture, or a laboratory period for any reason within the first two weeks of class, you will be dropped from the course.

**Academic integrity :** Academic dishonesty is a serious offense. Students are also expected to abide by the Academic Integrity policy of De Anza college. Details can be found at, <http://www.deanza.edu/studenthandbook/academic-integrity.html>. Copying another student’s data, paper, exam, quiz or use of technology devices to exchange information during class time and/or testing is never tolerated and result in **dismissal** from the course with **Grade F**.

**Cell Phone Policy:** Use of cell phone during lecture and lab sessions are strictly prohibited. Violation of this policy will bar you from attending the classes and may result in failure in the class.

**Homework :** Working problems at the end of each chapter is the assured way to increase your understanding of the course material. As this is a college-level course, homework will not be collected or graded; it is entirely up to you to discipline yourself to do as many problems as may be necessary for you. The suggested problems are **not** necessarily an indicator of the types of problems that will be found on quizzes or exams. Recommended problems are posted below.

Chapters	Problems
Ch:1- Matter and Measurement	1.19, 1.23, 1.25, 1.28, 1.30, 1.33, 1.35, 1.37, 1.41, 1.44, 1.47, 1.51, 1.55,1.56, 1.59, 1.67, 1.69, 1.70, 1.71.
Ch:2- Atoms and the Periodic Table	2.19, 2.22, 2.23, 2.25, 2.29, 2.31, 2.33, 2.36, 2.39, 2.41, 2.44, 2.45, 2.47, 2.53, 2.56, 2.57, 2.59, 2.61, 2.64,2.65, 2.69, 2.73, 2.76.
Ch:3-Ionic Compounds	3.17, 3.21, 3.24, 3.27, 3.29, 3.31, 3.33, 3.35, 3.36, 3.47, 3.41, 3.43, 3.47, 3.48, 3.53, 3.59, 3.61, 3.65, 3.67, 3.71, 3.74, 3.75, 3.77.
Ch:4- Covalent Compounds	4.16, 4.17, 4.19, 4.23, 4.25, 4.28, 4.31, 4.33, 4.35, 4.38 4.41, 4.43, 4.45, 4.48, 4.49, 4.54, 4.59, 4.61, 4.63, 4.68, 4.71, 4.77.
Ch:5-Chemical Reactions	5.23, 5.25, 5.28, 5.31, 5.35, 5.37, 5.41, 5.43, 5.45, 5.47, 5.50, 5.53, 5.55, 5.57, 5.62, 5.65, 5.67, 5.71, 5.75, 5.77, 5.79, 5.81, 5.85, 5.89.
Ch:6- Energy Changes, Reaction Rates, and Equilibrium.	6.11, 6.13, 6.16, 6.17, 6.21, 6.23, 6.25, 6.27, 6.29, 6.31, 6.33, 6.35, 6.40-44, 6.47, 6.49, 6.51, 6.53, 6.55, 6.57, 6.63, 6.65-67, 6.68-71.
Ch:7- Gases, Liquids, and Solids	7.19, 7.25, 7.28 – 31, 7.35, 7.39, 7.43, 7.46, 7.45 – 7.47, 7.57, 7.59, 7.61,7.63, 7.65, 7.69 – 73, 7.75, 7.77, 7.79 – 91.
Ch:8-Solutions	8: 23, 8.25, 8.27,8.29, 8.30, 8.33, 8.37, 8.41, 8.43, 8.44, 8.47, 8.50, 8.53, 8.59, 8.61, 8.65, 8.71, 8.73, 8.75,8.78, 8.83, 8.85, 8. 87.
Ch:9-Acids and Bases	9.21,9.23, 9.25, 9.27, 9.29, 9.33, 9.35,9.37, 9.39, 9.41, 9.43, 9.45, 9.47, 9.49, 9.52, 9.55, 9.59, 9.61, 9.63, 9.65, 9.67, 9.69, 9.71, 9.73,9.77, 9.81, 9.83, 9.85, 9.87, 9.90-91, 9.93, 9.95, 9.97, 9.100.
Ch:10-Nuclear Chemistry	10.19, 10.21, 10.23, 10.25, 10.27, 10.29, 10.31, 10.33, 10.35, 10.37, 10.39, 10.41, 10.43, 10.45, 10.47, 10.51, 10.53, 10.56,10.59, 10.61, 10.63, 10.65,10.67, 10.69.

**Chemical Disposal:** As a concern for the environment, proper chemical disposal is essential. Students who do not comply with directed procedures may be dropped from the course for repeated offenses.

**Eye protection:** You must wear full goggles that are sold by the De Anza Bookstore only and not safety glasses. Without them, you may not participate in lab and will receive a grade of zero for that lab.

**Changes to Syllabus:** This syllabus may change according to the needs of the class. Please check with the syllabus posted.

### Tentative Laboratory, Lecture, and Exam Schedule

Date Monday	Lecture(Rm-S35) <b>Lab-Section-01(Rm-SC2204)</b>	Date Wednesday	Lecture(Rm-S35) <b>Lab-Section-02(Rm-SC2204)</b>
08 Apr	Lecture :Introduction Ch.1: Matter and Measurement <b>Section-01-Lab: Check-In</b>	10 Apr	Lecture: Ch.2: Atoms and Periodic Table <b>Section-02-Lab: Check-In</b>
15 Apr	Lecture :Ch.2: cont. <b>Quiz 1: Ch. 1</b> <b>Section 01-Lab 1 : Measurements</b> <b>Signed Safety Document due</b>	17 Apr	Lecture:Ch.3: Ionic Compounds, <b>Quiz-2:Ch.2</b> <b>Section-02 -Lab 1 : Measurements</b> <b>Signed Safety Document due</b>
22 Apr	Lecture: Ch.3. Cont.,Ch. 4: Covalent Compounds <b>Section 01-Lab 2 : Nomenclature</b>	24 Apr	Lecture :Ch:4 cont, <b>Quiz 3: Ch.3,</b> <b>Section 02-Lab 2 : Nomenclature</b>
29 Apr	Lecture: Review for Exam-1(Ch.1-4) <b>Quiz-4: Ch.4 ; Section 01-Lab 3 : Models</b>	01 May	Lecture: <b>Exam-1(Ch.1-4)</b> <b>Section 02-Lab 3 : Models</b>
06 May	Lecture :Ch.5: Chemical Reactions <b>Section 01-Lab-4 : Hydrates (Part-1)</b>	08 May	Lecture:Ch.5: cont, Ch:6: Energy Changes, Reaction Rates, and Equilibrium <b>Section 02-Lab-4 : Hydrates(Part-1)</b>
13 May	Lecture: Ch: 6: cont., <b>Quiz 5: Ch. 5</b> Ch:7 Gases, Liquids and Solids <b>Section 01-Lab-5 : Hydrates(Part-2)</b>	15 May	Lecture:Ch.7: cont. <b>Quiz-6: Ch.6</b> <b>Section 02-Lab-5 : Hydrates(Part-2)</b>
20 May	Lecture: Ch:8:Solutions <b>Section 01-Lab-6 : Molar Volume</b>	22 May	Lecture:Ch:8. Contd, Review for Exam-2 (Ch.5-7) <b>Quiz-7: Ch.7</b> <b>Section 02-Lab-6 : Molar Volume</b>
27 May	<b>Memorial day, No classes</b>	29 May	Lecture: <b>Exam-2 (Ch.5-7)</b> <b>Section 02-Lab-7:Conductivity(Vernier)</b>
03 Jun	Lecture:Ch:9:Acids and Bases <b>Section 01-Lab-7:Conductivity(Vernier)</b>	05 Jun	Lecture:Ch:9:contd. <b>Quiz-8: Ch. 8</b> <b>.Section 02-Lab-8:Acid-Base Titration(Part-1)</b>
10 Jun	Lecture:Ch.10: Nuclear Chemistry <b>Section 01-Lab-8:Acid-Base Titration (Parts 1&amp;2)</b>	12 Jun	Lecture: Ch.10. Contd. <b>Quiz-9: Ch. 9</b> <b>Section 02-Lab-8: Acid-Base Titration(Part-2)</b>
17 Jun	Lecture :Review for Exam-3(Ch.8-10) Review for Final(Ch.1-10) <b>Quiz 10 : Ch-1-10</b> <b>Section 01 Lab- Lab Final ; Check-Out</b>	19 Jun	<b>Lecture: Exam -3(Ch.8-10)</b> <b>Section 02 Lab - Lab Final ; Check-Out</b>
24 Jun		26 Jun	<b>Final exam 1:45 PM to 3:45 PM</b>

**Lecture Final Exam on June 26 Wednesday, 1:45 PM to 3:45 PM**

## Laboratory Safety Rules

From the American Chemical Society Safety In Academic Laboratories Guidelines, 7th Ed., the following mandatory minimum safety requirements must be followed by all students and be rigorously enforced by all Chemistry faculty:

- 1) Chemistry Department-approved safety goggles purchased from the De Anza College bookstore (NOT safety glasses) must be worn at all times once laboratory work begins, including when obtaining equipment from the stockroom or removing equipment from student drawers, and may not be removed until all laboratory work has ended and all glassware has been returned to student drawers.
- 2) Shoes that completely enclose the foot are to be worn at all times; NO sandals, open-toed, or open-topped shoes, or slippers, even with socks on, are to be worn in the lab.
- 3) Shorts, cut-offs, skirts or pants exposing skin above the ankle, and sleeveless tops may not be worn in the lab: ankle-length clothing must be worn at all times.
- 4) Hair reaching the top of the shoulders must be tied back securely.
- 5) Loose clothing must be constrained.
- 6) Wearing jewelry such as rings, bracelets, and wristwatches in the laboratory should be discouraged to prevent chemical seepage in between the jewelry and skin.
- 7) Eating, drinking, or applying cosmetics in the laboratory is forbidden at ALL times, including during lab lecture.
- 8) Use of electronic devices requiring headphones in the laboratory is prohibited at ALL times, including during lab lecture.
- 9) Students are advised to inform their instructor about any pre-existing medical conditions, such as pregnancy, epilepsy, or diabetes, that they have that might affect their performance.
- 10) Students are required to know the locations of the eyewash stations, emergency shower, and all exits.
- 11) Students may not be in the lab without an instructor being present.
- 12) Students not enrolled in the laboratory class may not be in the lab at any time after the first lab period of each quarter.
- 13) Except for soapy or clear rinse water from washing glassware, NO CHEMICALS MAY BE Poured INTO THE SINKS; all remaining chemicals from an experiment must be poured into the waste bottle provided.
- 14) Students are required to follow the De Anza College Code of Conduct at all times while in lab: "horseplay", yelling, offensive language, or any behavior that could startle or frighten another student is not allowed during lab;
- 15) Strongly recommended: Wear Nitrile gloves while performing lab work; wear a chemically resistant lab coat or lab apron; wear shoes made of leather or polymeric leather substitute.

**Student Learning Outcome(s):**

\*Solve stoichiometric problems by applying appropriate molar relationships.

\*Identify the differences between elements and compounds and describe the chemical bonding in compounds- ionic vs. covalent.